

# 2-C-Branched Glycosides from 2'-Carbonylalkyl 2-O-Ms(Ts)-C-Glycosides. A Tandem $S_N2-S_N2$ Reaction via 1,2-Cyclopropanated Sugars

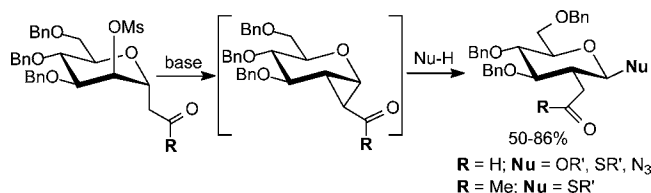
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## ABSTRACT



Under basic conditions, 2'-aldehyde (acetyl) 2-O-Ms(Ts)- $\alpha$ -C-glycosides undergo an intramolecular  $S_N2$  reaction to form 1,2-cyclopropanated sugars, which react with nucleophiles (alcohols, thiols, and azide) at the anomeric carbon to give 2-C-branched glycosides. By way of contrast, the 1,2-cyclopropanes derived from 2'-ketones only react with thiols to give 2-C-branched thioglycosides.

C-Branched sugars in natural antibiotics, bacterial polysaccharides, and macrolides are often associated with specific biological functions.<sup>1</sup> Unnatural 2-C-branched sugars also serve as metabolic substrates, e.g., Bertozzi et al. tested 2-C-acetonylsugars, derived from 2-iodosugars, as mimics of 2-N-acetylsugars for cell surface engineering<sup>2</sup> and Hindsgaul et al. prepared a 2-C-acetamide sugar from a 2,3-epoxide as an inhibitor of the biosynthesis of lipid A.<sup>3</sup>

Most 2-C-branched sugars are synthesized from glycals through 1,2-cyclopropanation followed by selective ring-opening via solvolysis,<sup>4,5</sup> which often provides an anomeric mixture of glycosides because of the involvement of an oxocarbenium-like intermediate, with  $\alpha$ -glycosides being

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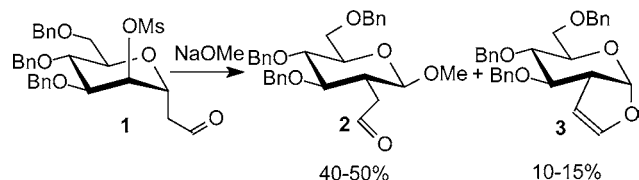
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### Scheme 1. Base-Mediated Rearrangements

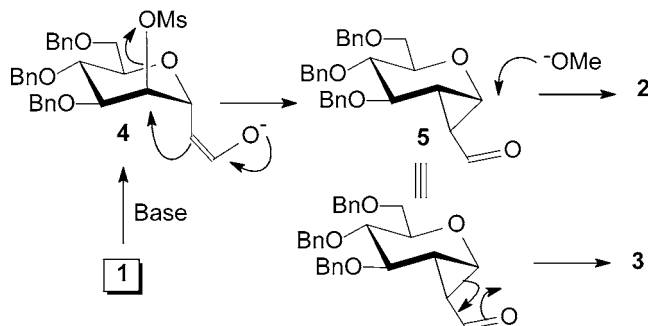


avored due to the anomeric effect. A recent ring-opening of sugar cyclopropanecarboxylates mediated by NIS provided 1,2-trans 2-C-branched glycosides.<sup>6</sup> We report herein a tandem  $S_N2-S_N2$  reaction involving a base-mediated 1,2-cyclopropanation from 2'-carbonylalkyl 2-O-Ms(Ts)-C-glycosides and subsequent nucleophilic substitution at the anomeric carbon leading to 1,2-trans 2-C-branched  $\beta$ -O- and  $\beta$ -S-glycosides and  $\beta$ -glycosyl azides.

It is known that 2'-carbonylalkyl  $\alpha$ -C-glycosides epimerize to their  $\beta$ -anomers under basic conditions through  $\beta$ -elimination to an acyclic  $\alpha,\beta$ -conjugated aldehyde (ketone) intermediate followed by an intramolecular hetero-Michael addition.<sup>7</sup> However, this cyclization is poorly stereoselective in C-furanosides. Fleet et al. have developed an intramolecular  $S_N2$  reaction to form furan esters via base treatment of 2-O-Tf (Ms) sugar lactones.<sup>8</sup> Following the same rationale, we decided to place a leaving group at the O(2)-position of 2'-carbonylalkyl C-glycopyranosides to see if the C-furanoside would be formed stereoselectively after  $\beta$ -elimination and subsequent intramolecular  $S_N2$  substitution at C(2) by C(5)-OH.

Thus, 2'-aldehyde 2-O-Ms-C-glycoside (**1**) was prepared from the respective allyl C-glycoside by ozonolysis.<sup>9</sup> Upon treatment of **1** with 4% NaOMe, two major products were obtained, namely, the 2-C-branched methyl  $\beta$ -glucopyranoside **2** (40–50%) and the bicyclic product **3** (10–15%), but no C-furanoside was isolated (Scheme 1). Both structures were unambiguously characterized by NMR analysis.<sup>10</sup> On the basis of the products obtained, we believe that the enolate **4** reacted to give the 1,2-cyclopropanated sugar **5**, which in turn underwent ring-opening with methoxide at the anomeric carbon to afford **2**; presumably an intramolecular rearrangement afforded **3** (Scheme 2).

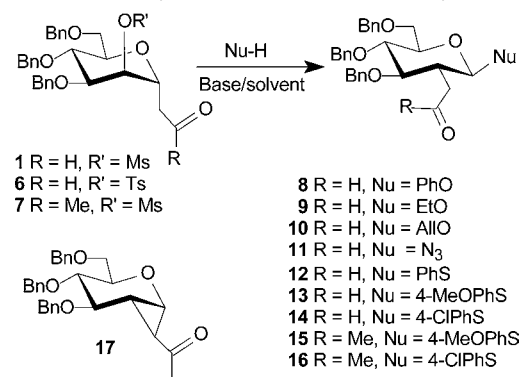
### Scheme 2 Tandem $S_N2-S_N2$ Mechanism



The results indicate that the cyclopropanation (1,2-substitution) was favored over the  $\beta$ -elimination due to the 1,2-trans configurations of  $\alpha$ -C-mannoside. This mechanism resembles one observed by Danishefsky et al. involving 1,2-migration of the *N*-sulfonamide from 2-iodo-1-*N*-sulfonamide to 2-*N*-sulfonamide glycosyl compounds.<sup>11</sup>

Although highly  $\beta$ -selective, the reaction will not be practically useful unless better chemical selectivity is achieved and other substrates and nucleophiles can be incorporated. Thus, 2'-aldehyde 2-O-Ts-C-glycoside (**6**) and 2'-acetylonyl 2-O-Ms-C-glycoside (**7**) were included as substrates, and alcohols, thiols, and sodium azide were used as nucleophiles. After examining various base/solvent combinations, we were able to obtain 2-C-branched glycosides (**2**, **9**, and **10**) from 2-O-Ms **1** in good yields with triethylamine as base, and no byproduct **3** was isolated (see Scheme 3 and entries 1, 4,

### Scheme 3. Synthesis of 2-C-Branched Glycosides



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(10)  $\beta$ -Configuration of **2** was determined on the basis of the observation of NOE between H1 and H3 and the large coupling constant  $J_{1,2} = 8.8$  Hz.

and **6** in Table 1). The same selectivity but lower yields were obtained from 2-O-Ts **6** with  $K_2CO_3/CH_3CN$  as the base and solvent (entries 2, 3, and 5). Under both sets of conditions, the 2-C-branched glycosyl azide **11** was also produced in moderate yield (entry 7). The best results, however, were obtained when **1**, **6**, and **7** were treated with thiols; this

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**Table 1.** Synthesis of 2-C-Branched Glycosides<sup>a</sup>

entry	substrate	nucleophile	base/solvent <sup>b</sup>	product (yield %)
1	<b>1</b>	MeOH	A	<b>2</b> (72%)
2	<b>6</b>	MeOH	B	<b>2</b> (52%)
3	<b>6</b>	PhOH	B	<b>8</b> (62%)
4	<b>1</b>	EtOH	C	<b>9</b> (71%)
5	<b>6</b>	EtOH	B	<b>9</b> (51%)
6	<b>6</b>	AllylOH	A	<b>10</b> (76%)
7	<b>1</b> and <b>6</b>	NaN <sub>3</sub>	A and B	<b>11</b> (50–52%)
8	<b>1</b> and <b>6</b>	PhSH	B and D	<b>12</b> (67–86%)
9	<b>1</b> and <b>6</b>	4-MeOPhSH	B and D	<b>13</b> (70–85%)
10	<b>1</b> and <b>6</b>	4-ClPhSH	B and D	<b>14</b> (72–83%)
11	<b>7</b>	4-MeOPhSH	D	<b>15</b> (83%)
12	<b>7</b>	4-ClPhSH	D	<b>16</b> (85%)
13	<b>7</b>	MeOH	D	<b>17</b> (81%)

<sup>a</sup> Reaction was conducted at room temperature overnight. <sup>b</sup> Base/solvent combinations: A, TEA/MeOH; B, K<sub>2</sub>CO<sub>3</sub>/MeCN; C, TEA/EtOH; D, K<sub>2</sub>CO<sub>3</sub>/MeOH.

produced 2-C-branched  $\beta$ -thioglycosides (**12–16**) in excellent yields (entries 8–12). Surprisingly, when **7** was treated with base in methanol, rather than the 2-C-branched methyl  $\beta$ -glycoside being isolated, we obtained the stable 1,2-cyclopropanated sugar **17**, an intermediate predicted by the mechanism illustrated in Scheme 2, as the major product (entry 13).<sup>12</sup> Treatment of **17** with thiols in K<sub>2</sub>CO<sub>3</sub>/MeOH produced the expected 2-C-branched thioglycosides, but its reaction with alcohols (MeOH and PhOH) was very sluggish and yielded multiple byproducts. The diminished reactivity

(12) An attempt to isolate 1,2-cyclopropanated **5** was unsuccessful. For **17**:  $\delta_{\text{H}}$  1.98 ppm (H-2,  $J_{1,2}$  = 7.2 Hz), 2.34 (H-1',  $J_{1,1'}$  = 1.6), and 3.86 (H-1,  $J_{2,1'}$  = 5.6 Hz). For complete assignment, see Supporting Information.

of **17** in comparison to **5** is probably due to the acetyl group being less electron-withdrawing than the aldehyde.<sup>4e,13</sup> Consequently, cyclopropane **17** only reacted with strong nucleophiles. To the best of our knowledge, these are the first examples of an S<sub>N</sub>2 reaction at the anomeric carbon of 1,2-cyclopropanated sugars.

In summary, we have developed an alternative method for the synthesis of 2-C-branched glucosides with exclusive  $\beta$ -anomeric selection. 2'-Aldehyde (acetonyl) 2-O-Ms(Ts)- $\alpha$ -C-mannosides undergo an intramolecular S<sub>N</sub>2 reaction to form 1,2-cyclopropanated sugars. The cyclopropane intermediates from the 2'-aldehydes then react with nucleophiles (alcohols, thiols, and azide) by an S<sub>N</sub>2 reaction at the anomeric carbon leading to 2-C-branched glycosides, while 1,2-cyclopropanes derived from 2'-ketones only react with thiols to give 2-C-branched thioglycosides. Further studies on the application of this reaction and the utilities of the intermediates are currently in progress.

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**Supporting Information Available:** Experimental procedures and NMR spectra (<sup>1</sup>H, <sup>13</sup>C, COSY, NOESY, TOCSY, and HSQC) for products (**2**, **3**, **6–17**). This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(13) Nucleophilic addition only takes place on cyclopropanes substituted with electron-withdrawing groups, see: Danishefsky, S. J. *Acc. Chem. Res.* **1979**, *12*, 66–72.